

**UNCLASSIFIED**

AD **408 322**

**DEFENSE DOCUMENTATION CENTER**

FOR

**SCIENTIFIC AND TECHNICAL INFORMATION**

CAMERON STATION, ALEXANDRIA, VIRGINIA



**UNCLASSIFIED**

NOTICE: When government or other drawings, specifications or other data are used for any purpose other than in connection with a definitely related government procurement operation, the U. S. Government thereby incurs no responsibility, nor any obligation whatsoever; and the fact that the Government may have formulated, furnished, or in any way supplied the said drawings, specifications, or other data is not to be regarded by implication or otherwise as in any manner licensing the holder or any other person or corporation, or conveying any rights or permission to manufacture, use or sell any patented invention that may in any way be related thereto.

## The Effect of an Electric Field on the Critical Opalescence.

### 1. Introduction.

Concentration fluctuations in liquid mixtures, which may be observed by means of light scattering, include also fluctuations of the dielectric constant. If an electric field is applied, the fluctuations of the dielectric constant cause changes of the electric energy density, which is connected with the dielectric constant and the field strength by the relation

$$W = \frac{\epsilon}{\epsilon_0} E^2 \quad (1)$$

In the expression for the free energy of a binary liquid system an additional term due to the electric energy has to be added to the term representing the work involved in concentration changes. The work necessary to create a given amplitude of concentration fluctuations is usually rather big, and it is only in the vicinity of the critical point that it becomes small enough to become comparable to the electric work involved.

In calculating the possible effect of an electric field on the concentration fluctuations one has to consider, that both the dielectric constant and the field strength are functions of the geometrical distribution of the concentration fluctuations.

### 2. Concentration Fluctuations and Electric Energy.

The energy density of a parallel plate condenser is

given by

$$w = \frac{\epsilon}{8\pi} E^2$$

$\epsilon$  is the dielectric constant of the medium and  $E$  is the field strength of the applied electric field.

Concentration fluctuations create fluctuations of the dielectric constant. Observing the fluctuations by means of light scattering and accepting Brillouins interpretation of the fluctuations as a superposition of sonic waves, we may write for one of the components

$$\epsilon = \epsilon_0 + \eta \sin K(\bar{S}, \bar{r}) \quad (2)$$

The vector  $\bar{r}$  defines the position of the point in which the dielectric constant is measured. The unit vector  $\bar{S}$  has the direction of propagation of the sound wave.  $K = \frac{2\pi}{\Lambda}$  in which  $\Lambda$  is the wavelength of the wave and  $\eta$  is the amplitude of the fluctuation of the dielectric constant.

The field strength may be expressed as the gradient of a potential  $\Phi$ :  $E = -\text{grad } \Phi$  This potential  $\Phi$  is a function of the position defined by the vector  $\bar{r}$ . We develop it in powers of  $\eta$  and obtain

$$\Phi = \Phi_0 + \eta \Phi_1 + \eta^2 \Phi_2 \quad (3)$$

For an electrostatic field we always have the relation

$$\text{div}(\epsilon E) = 0$$

Substituting from eq. (2) and  $\Phi$  from eq. (3) we obtain after arrangement according to powers of  $\eta$ .

$$\text{div} \left\{ \begin{array}{l} \epsilon_0 \text{grad} \Phi_0 + \eta \epsilon_0 \text{grad} \Phi_1 + \eta^2 \epsilon_0 \text{grad} \Phi_2 \\ + \eta \sin K(\bar{S}, \bar{r}) \text{grad} \Phi_0 + \eta^2 \sin K(\bar{S}, \bar{r}) \text{grad} \Phi_1 \end{array} \right\} = 0$$

neglecting all terms containing higher powers of  $\eta$  than the second. So the equations necessary to find  $\Phi_0, \Phi_1, \Phi_2$  are

$$\Delta \Phi_0 = 0 \quad (4a)$$

$$(4) \quad \Delta \Phi_1 = -\frac{1}{\epsilon} \operatorname{div} [\sin K(\bar{s}, \bar{\pi}) \operatorname{grad} \Phi_0] \quad (4b)$$

$$\Delta \Phi_2 = -\frac{1}{\epsilon_0} \operatorname{div} [\sin K(\bar{s}, \bar{\pi}) \operatorname{grad} \Phi_1] \quad (4c)$$

We now assume the electric field to be applied in the  $z$ -direction. Then we may write..

$$(5) \quad \Phi_0 = -A z$$

Now calculating  $\operatorname{grad} \Phi_0$  from equ. (4a) and substituting this value in equ. (4b) we obtain  $\Delta \Phi_1$  and  $\operatorname{grad} \Phi_1$ , and repeating this operation with  $\Phi_1$  we also obtain  $\operatorname{grad} \Phi_2$ .

First we have

$$\operatorname{grad}_x \Phi_0 = 0, \quad \operatorname{grad}_y \Phi_0 = 0, \quad \operatorname{grad}_z \Phi_0 = -A$$

which gives

$$\Delta \Phi_1 = \frac{A}{\epsilon_0} \frac{\partial}{\partial z} [\sin K(\bar{s}, \bar{\pi})] = \frac{A}{\epsilon_0} K \gamma \cos K(\bar{s}, \bar{\pi})$$

supposing that the 3 components of the unit-vector  $\bar{s}$  are  $\alpha, \beta, \gamma$ .

From this it follows that we can take for  $\Phi_1$  the expression

$$\Phi_1 = -\frac{A}{\epsilon_0 K} \gamma \cos K(\bar{s}, \bar{\pi})$$

which yields

$$\operatorname{grad}_x \Phi_1 = \frac{A}{\epsilon_0} \gamma \alpha \sin K(\bar{s}, \bar{\pi})$$

$$\operatorname{grad}_y \Phi_1 = \frac{A}{\epsilon_0} \gamma \beta \sin K(\bar{s}, \bar{\pi})$$

$$\operatorname{grad}_z \Phi_1 = \frac{A}{\epsilon_0} \gamma^2 \sin K(\bar{s}, \bar{\pi})$$

Proceeding now to the calculation of  $\Phi_2$  we have

$$\Delta \Phi_2 = -\frac{A}{\epsilon_0^2} \gamma (\alpha \frac{\partial}{\partial x} + \beta \frac{\partial}{\partial y} + \gamma \frac{\partial}{\partial z}) \sin^2 K(\bar{s}, \bar{\pi})$$

$$\text{or} \quad \Delta \Phi_2 = -\frac{A}{\epsilon_0^2} K \gamma \sin 2K(\bar{s}, \bar{\pi})$$

which gives

$$\tilde{\phi}_2 = \frac{A\gamma}{\epsilon_0} \frac{1}{4K} \sin 2K(\bar{s}, \bar{\pi})$$

So finally, in second approximation, we have

$$(6) \quad \tilde{\phi} = -A\bar{z} - \eta \frac{A\gamma}{\epsilon_0 K} \cos K(\bar{s}, \bar{\pi}) + \eta^2 \frac{A\gamma}{4\epsilon_0^2 K} \sin 2K(\bar{s}, \bar{\pi})$$

From this expression for the potential we derive the field-strength  $E$  by calculating the 3 components of the gradient and obtain

$$E_x = -\eta \frac{A\gamma}{\epsilon_0} \alpha \sin K(\bar{s}, \bar{\pi}) - \eta^2 \frac{A\gamma}{2\epsilon_0^2} \alpha \cos 2K(\bar{s}, \bar{\pi})$$

$$(6') \quad E_y = -\eta \frac{A\gamma}{\epsilon_0} \beta \sin K(\bar{s}, \bar{\pi}) - \eta^2 \frac{A\gamma}{2\epsilon_0^2} \beta \cos 2K(\bar{s}, \bar{\pi})$$

$$E_z = A - \eta \frac{A\gamma}{\epsilon_0} \gamma \sin K(\bar{s}, \bar{\pi}) - \eta^2 \frac{A\gamma}{2\epsilon_0^2} \gamma \cos 2K(\bar{s}, \bar{\pi})$$

Finally we have to calculate the energy-density

$$W = \frac{\epsilon}{8\pi} E^2$$

by substituting for  $E$  expressions from eq.(2) and for the 3 components of  $E$  their expressions from (6'). The result up to the second power of  $\eta$  is

$$8\pi W = \epsilon_0 A^2 - \frac{\eta^2}{\epsilon_0} A^2 \gamma^2 \sin^2 K(\bar{s}, \bar{\pi}) + \text{terms which contain only first powers of the periodic functions.}$$

Since the phase of the wave must be arbitrary we can average over the periodic function and obtain

$$(7) \quad W = \frac{\epsilon_0 A^2}{8\pi} \left[ 1 - \frac{1}{2} \left( \frac{\eta}{\epsilon_0} \right)^2 \gamma^2 \right]$$

This means that in such a fluctuation only the component of the applied electrical field in the direction of propagation of the supersonic wave is efficient.

### 3. Critical opalescence in an electric field

For a binary liquid mixture of two components 1 and 2 occupying a volume  $V$  in which the fluctuations of the volume fraction  $\varphi_2$  of component 2 have an amplitude  $\zeta$ , the free

energy  $F$ , as changed by those fluctuations, can be represented by the expression

$$(8) \quad \frac{F}{VkT} = \frac{F_0}{VkT} + \frac{1}{4} \left( \frac{\gamma}{\varphi_2} \right)^2 \varphi_2 \frac{\partial}{\partial \varphi_2} \left( \frac{P}{kT} \right)$$

This holds as long as there is no external field and was derived in the article on the "Angular Dissymmetry of the Critical Opalescence in Liquid Mixtures" J. Chem. Phys. 31, 680 (1959). It is equation (24) of that article in which the dissymmetry-effect has been neglected. This is sufficiently correct for the estimate we are going to make of the expected order of magnitude of the electrical field effect.

It is obvious that in the presence of a field  $A$  we will have to add to the right-hand side of equation (8) the additional electrical energy calculated in eq.(7) divided by  $kT$ . In this way we obtain

$$(8') \quad \frac{F}{VkT} = \frac{F_0}{VkT} + \frac{1}{4} \left( \frac{\gamma}{\varphi_2} \right)^2 \varphi_2 \frac{\partial}{\partial \varphi_2} \left( \frac{P}{kT} \right) - \frac{\epsilon_0 A^2 \gamma^2}{16\pi kT} \left( \frac{\eta}{\epsilon_0} \right)^2$$

We substitute the concentration  $c$  for the volume fraction of the second component. Since

$$\frac{\eta}{\epsilon} = \frac{\gamma}{\varphi} \left( \frac{\epsilon}{\epsilon_0} \frac{\partial \epsilon}{\partial \varphi} \right) = \frac{\gamma}{\varphi} \left( \frac{c}{\epsilon_0} \frac{\partial \epsilon}{\partial c} \right)$$

we now can write instead of eq. (8')

$$(8'') \quad \frac{F}{VkT} = \frac{F_0}{VkT} + \frac{1}{4} \left( \frac{\gamma}{\varphi_2} \right)^2 \left[ c \frac{\partial}{\partial c} \left( \frac{P}{kT} \right) - \frac{\epsilon_0 A^2 \gamma^2}{4\pi kT} \left( \frac{c}{\epsilon_0} \frac{\partial \epsilon}{\partial c} \right)^2 \right]$$

In the case  $A=0$  eq (25) of the paper in the J.Chem. Phys. loc. cit. gives the scattered intensity. It is clear that when  $A \neq 0$  we will have to substitute for the denominator in this formula the expression in bracket of eq.(8'')

Suppose now that the experiment is performed at the critical concentration. In this case it follows from equations (22) (22') and (26) in J.Chem. Phys. loc cit. that

$$c \frac{\partial}{\partial c} \left( \frac{P}{kT} \right) = \frac{1}{\omega_2} \left( \frac{T}{T_c} - 1 \right)$$

in which  $\omega_2$  is the volume occupied by one molecule of the second component. Since the reciprocal scattered intensity is proportional to the expression in brackets of eq. (8'') it follows in essence that we can expect an observable effect by the application of a field  $A$  if we can make the electric energy content of a volume equal to that occupied by one molecule of the second component of the same order as

$$kT \left( \frac{T}{T_c} - 1 \right)$$

The chances of an observable effect therefore increase with diminishing temperature distance of the critical temperature. At the same time the minus-sign in the bracket indicates that under the influence of an applied field the scattered intensity should increase.

This can be expressed in a quantitative way by comparing the intensity for  $A=0$  to the intensity after application of the field. It follows that

$$(9) \frac{I_{A=0}}{I_{A=A}} = \frac{\frac{\Delta T}{T_c} - \left( \frac{c}{\epsilon} \frac{\partial \epsilon}{\partial a} \right)^2 \frac{\omega_2 \epsilon A^2 r^2}{4\pi} / kT}{\Delta T / T_c}$$

if we call the temperature distance  $T - T_c = \Delta T$ . Another feature of the expected effect is that it is proportional to  $A^2 r^2$ , which means that only the component of the field perpendicular to the front of the reflecting supersonic wave is effective.

#### 4 Example (Methanol in Cyclohexane)

The dielectric constants of cyclohexane and methanol are

$\epsilon_1 = 2.0$  and  $\epsilon_2 = 32.6$ . The molar volumes are 107 and 40.5 cc, making  $\omega_1 = 177 \cdot 10^{-24}$  and  $\omega_2 = 67.3 \cdot 10^{-24}$  cc. This makes

$x = 0.37$  and the critical molfraction

$$(\varphi_2)_{\text{crit}} = \frac{1}{1+x} = 0.62$$

Assuming linear dependences of the dielectric constant on the volume fraction, which should be good enough for our immediate purpose, the dielectric constant at the critical concentration comes out to be 21.0 and

$$\left(\frac{c}{\epsilon} \frac{\partial \epsilon}{\partial c}\right)^2 = 0.81$$

If we now suppose the experiment of being performed by the observation of the scattered light under  $90^\circ$  with the primary beam and the field to be in the direction of the secondary beam we have  $\gamma = \frac{1}{\sqrt{2}}$  and for an applied field strength of

$30000 \frac{\text{Volts}}{\text{cm}} = 100 \text{ a.s. units}$  we obtain with  $k = 1.37 \cdot 10^{-16} \text{ ergs}$ :

$$\left(\frac{c}{\epsilon} \frac{\partial \epsilon}{\partial c}\right)^2 \frac{\omega_2 \epsilon A^2 r^2}{4\pi k} = 3.3 \cdot 10^{-3}$$

Since the temperature of observation will be very near the critical temperature this result has to be compared with the temperature-distance  $\Delta T$  from the critical temperature.

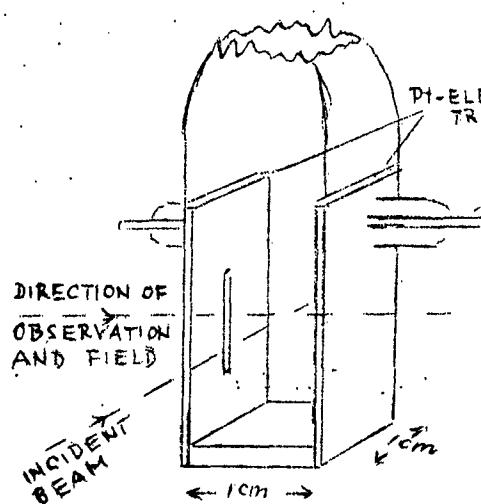
So if observation would be made at  $\Delta T = 0.03^\circ\text{C}$  the effect would be 11% of the primary intensity and 1.1% if the experiment is performed at  $\Delta T = 0.3^\circ\text{C}$ .

Examining different mixtures, which were all chosen to have as one component a polar substance (which obviously is helpful because of the relatively high dielectric constant of polar

substances) it turned out that all mixtures had relatively high electrical conductivities. The measured specific resistance was sometimes as low as  $10^8 \text{ ohm cm}$ . In this case the heating effect is rather large and it is estimated that under these circumstances the heating rate might be as high as  $5^\circ/\text{sec}$ . Obviously it is therefore necessary to apply a pulse technique which allows the field to be applied in short bursts of the order of 100 micro sec.

#### The experimental setup

The design and the dimensions of the measuring cell are shown in the figure.



It consists of a rectangular polished Pyrex cell, the molten in platinum electrodes are 1 cm apart from each other. The small slit through which the scattered light is observed by means of a photomultiplier should not disturb the homogeneity of the field.

The cell is fixed in a temperature bath which allows to control the temperature within  $.005^\circ$ . The refractive index of the oil used in the temperature bath is the same as the refractive index of the glass in order to avoid reflections. The electrodes are connected with the

high voltage power supply by heavily isolated cables. Temperature bath, measuring cell and phototube are placed in a black painted box, which is mounted on a optical bank, the light from a 500 Watt projection lamp enters through a system of slits. The photocurrent is measured by means of an oscilloscope.

As a power supply a high voltage pulse generator is used which delivers rectangular pulses up to 30 000 volts with a variable pulse width up from 100 microseconds.

TECHNICAL REPORT DISTRIBUTION LIST

<u>NO. COPIES</u>	<u>NO. COPIES</u>		
Commanding Officer Office of Naval Research Branch Office The John Crerar Library Building 86 East Randolph Street Chicago 1, Illinois	(1)	Air Force Office of Scientific Research (SRG-S) Washington 25, D. C.	(1)
Commanding Officer Office of Naval Research Branch Office 346 Broadway New York 13, New York	(1)	Commanding Officer Diamond Ordnance Fuze Laboratories Washington 25, D. C. Attn: Technical Information Office Branch 012	(1)
Commanding Officer Office of Naval Research Branch Office 1030 East Green Street Pasadena 1, California	(1)	Office, Chief of Research and Devel. Department of the Army Washington 25, D. C. Attn: Physical Sciences Division	(1)
Commanding Officer Office of Naval Research Branch Office Box 39 Navy No. 100 Fleet Post Office New York, New York	(7)	Chief, Bureau of Ships Department of the Navy Washington 25, D. C. Attn: Code 342C	(2)
Director, Naval Research Laboratory Washington 25, D. C. Attn: Technical Information Officer Chemistry Division	(6)	Chief, Bureau of Naval Weapons Department of the Navy Washington 25, D. C. Attn: Technical Library	(4)
Chief of Naval Research Department of the Navy Washington 25, D. C. Attn: Code 425	(2)	ASTIA Document Service Center Arlington Hall Station Arlington 12, Virginia	(10)
DDR and E Technical Library Room 3C-128, The Pentagon Washington 25, D. C.	(1)	Director of Research U. S. Army Signal Research and Devel. Laboratory Fort Monmouth, New Jersey	(1)
Technical Director Research and Engineering Division Office of the Quartermaster General Department of the Army Washington 25, D. C.	(1)	Naval Radiological Defense Laboratory San Francisco 24, California Attn: Technical Library	(1)
Research Director Clothing and Organic Materials Division Quartermaster Research and Engineering Command U. S. Army Natick, Massachusetts	(1)	Naval Ordnance Test Station China Lake, California Attn: Head, Chemistry Division	(1)

Distribution List.....page 2

<u>NO. COPIES</u>		<u>NO. COPIES</u>	
Commanding Officer Army Research Office Box CM, Duke Station Durham, North Carolina Attn: Scientific Synthesis Office	(1)	Dr. Sydney Ross Rensselaer Polytechnic Institute Troy, New York	(1)
Brookhaven National Laboratory Chemistry Department Upton, New York	(1)	Dr. U. P. Strauss Department of Chemistry Rutgers University New Brunswick, New Jersey	(1)
Atomic Energy Commission Division of Research Chemistry Programs Washington 25, D. C.	(1)	Dr. R. Ullman Department of Chemistry Polytechnic Institute of Brooklyn Brooklyn, New York	(1)
Atomic Energy Commission Division of Technical Information Extension Post Office Box 62 Oak Ridge, Tennessee	(1)	Dr. R. M. Fuoss Department of Chemistry Yale University New Haven, Connecticut	(1)
U. S. Army Chemical Research and Development Laboratories Technical Library Army Chemical Center, Maryland	(1)	Dr. E. Grunwald Department of Chemistry Florida State University Tallahassee, Florida	(1)
Office of Technical Services Department of Commerce Washington 25, D. C.	(1)	Commanding Officer and Director U. S. Naval Civil Engineering Laboratory Port Hueneme, California Attn: Chemistry Division	(1)
ONR Resident Representative University of Rochester Lattimore Hall, River Campus Rochester, New York	(1)	Dr. G. Barth-Wehrenalp, Director Inorganic Research Department Pennsalt Chemicals Corporation Post Office Box 4388 Philadelphia 18, Pennsylvania	(2)
Dr. C. R. Singletary Naval Research Laboratory Code 6170 Washington 25, D. C.	(1)	Dr. Blanton C. Duncan Surface Chemistry Section National Bureau of Standards Washington 25, D. C.	(1)
Dr. Karol J. Mysels Department of Chemistry University of Southern California Los Angeles 7, California	(1)	Monsanto Research Corporation Everett Station Boston 49, Massachusetts Attn: Mr. K. Warren Eagle	(1)
Dr. A. C. Zettlemoyer Department of Chemistry Lehigh University Bethlehem, Pennsylvania	(1)	Dr. M. S. Cohen Propellants Synthesis Section Reaction Motors Division Denville, New Jersey	(1)